



Modeling Atmospheric Chemistry and Aerosols

**Presented by Rebecca Buchholz,
Atmospheric Chemistry Observations & Modeling
(ACOM) Laboratory**

Slide Content Contributions:

*Simone Tilmes, Mike Mills,
Louisa Emmons, Doug Kinnison,
Kelley Barsanti, Wenfu Tang, Peter
Lawrence, Peter Lauritzen, Danny Leung*

Chemistry-Climate Working Group (CCWG)

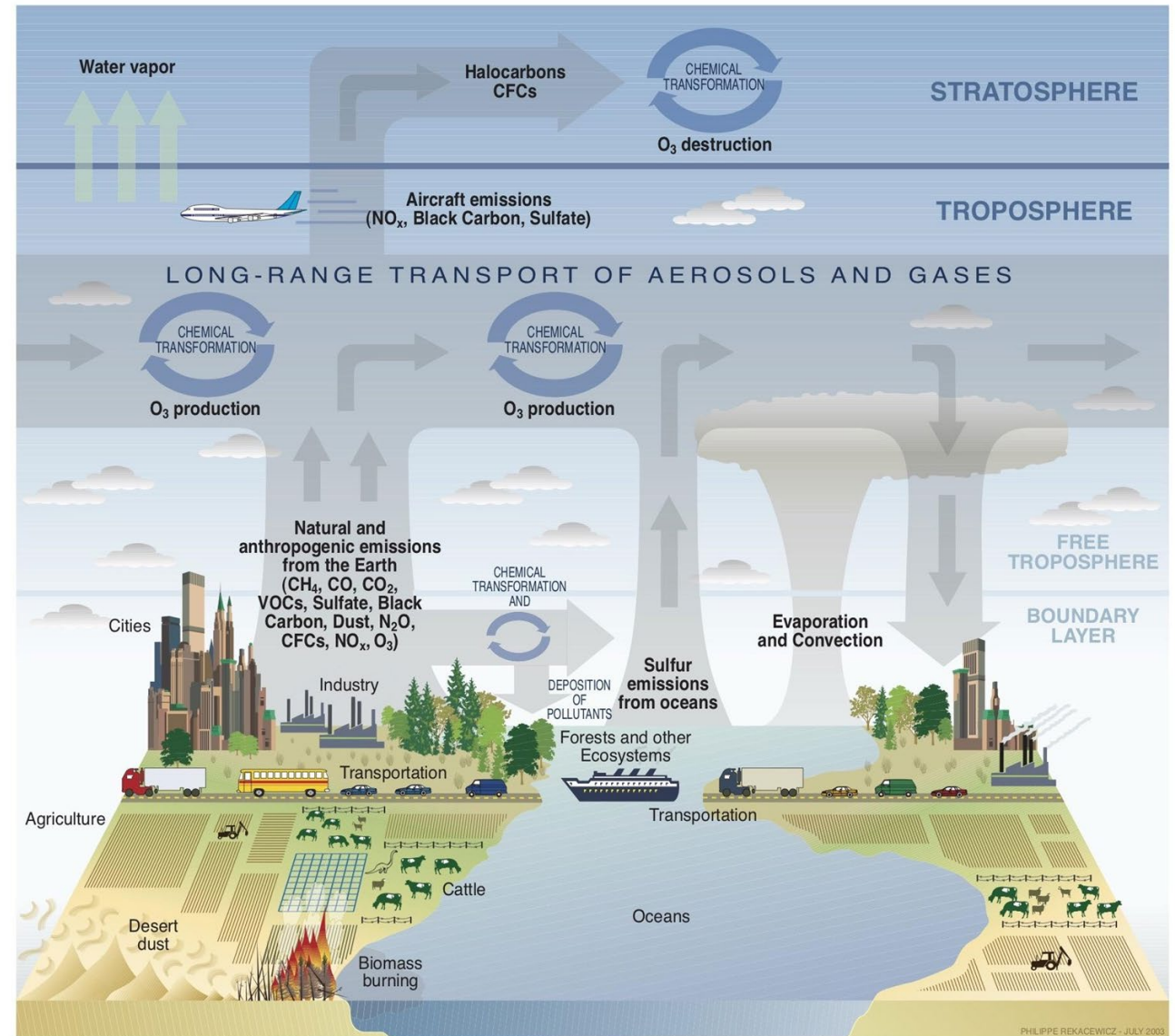
<i>CCWG Co-Chairs:</i>	<i>Simone Tilmes Rafael Fernandez</i>
<i>Software Engineer:</i>	<i>Francis Vitt</i>
<i>CCWG Liaisons:</i>	<i>Rebecca Buchholz</i>

July, 2025

Shawn Honomichl

Atmospheric Chemistry

- Motivation
- Adding processes into models
 - Emissions
 - Chemical mechanism
 - Aerosol model and cloud interactions
 - Dry Deposition
 - Wet Deposition
- Applications
- Support



Atmospheric Chemistry: Why is it important – Health

Ozone pollution (NO_x, CO, VOC, CH₄):

- Damages tissues, causes inflammation
- Coughing, chest tightness and worsening of asthma

Particulate Matter: PM_{2.5} and PM₁₀ diameter < 2.5 or 10 μm
(SO₂, VOC, NH₃, BC, OC, fine dust):

- Cardiovascular impacts (lungs and heart), premature deaths

Sources:

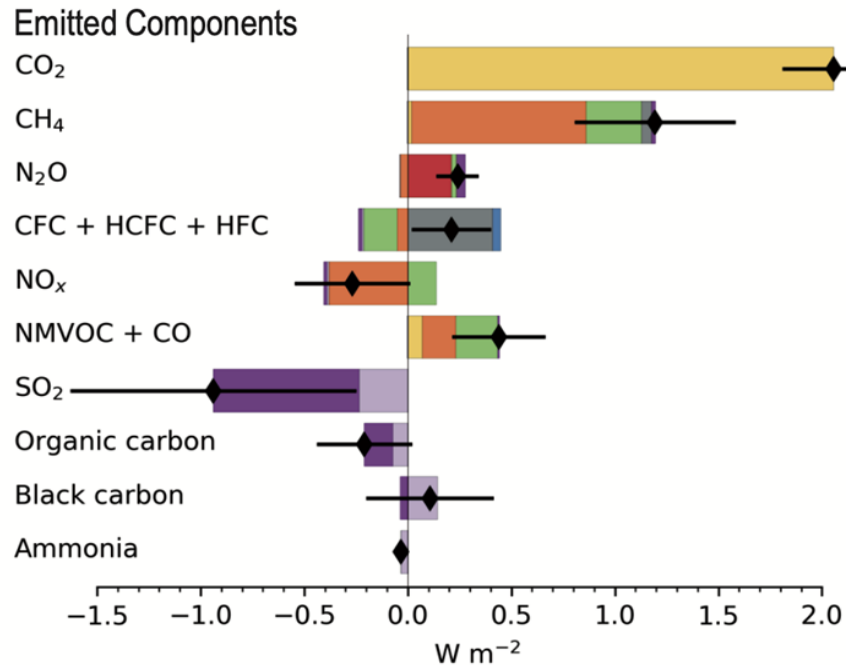
- Traffic / Industry & Private (use of fossil fuels)
- Farmland
- Fires
- Vegetation
- PM: Dust storms (worsen with climate change)
- PM: Volcanoes



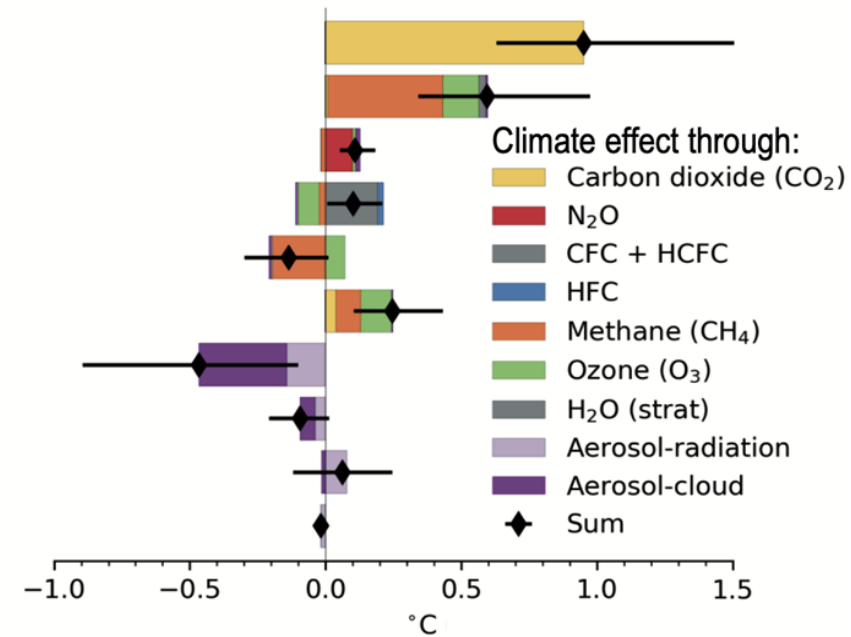
(7+ million premature deaths due to air pollution per year)

Atmospheric Chemistry: Why is it important – Climate

(a) Effective radiative forcing
1750 to 2019



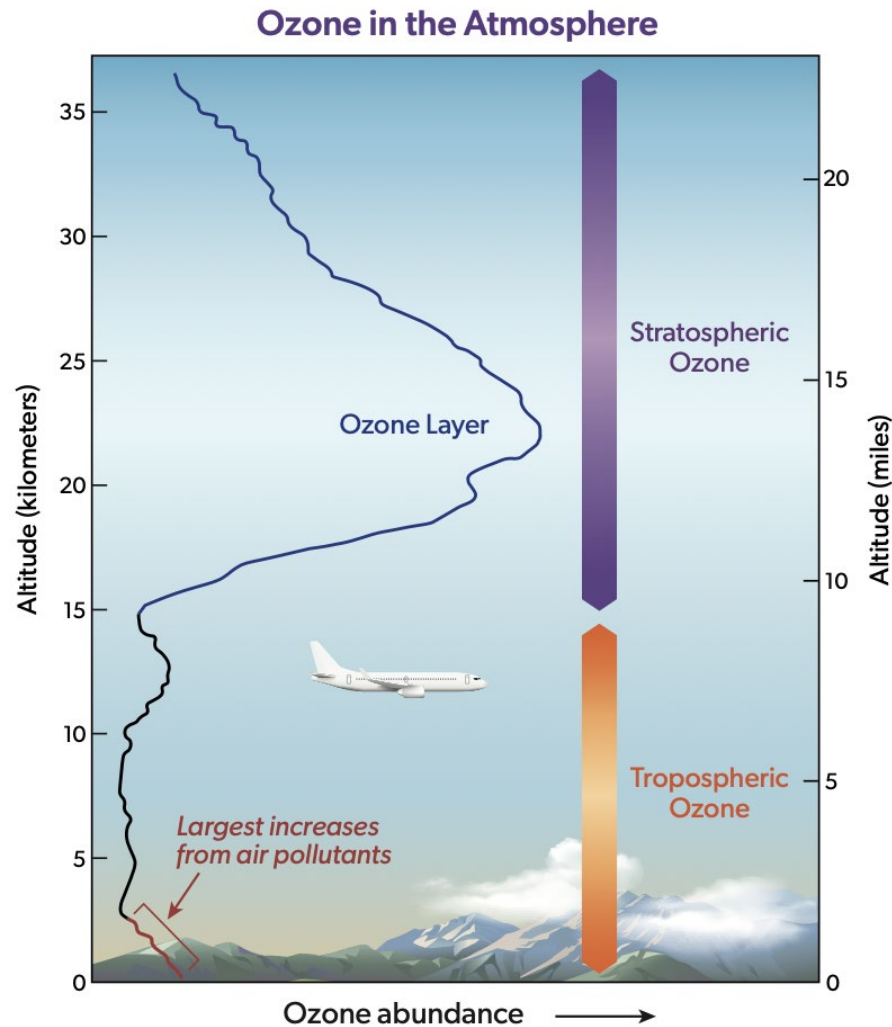
(b) Change in global surface temperature
1750 to 2019



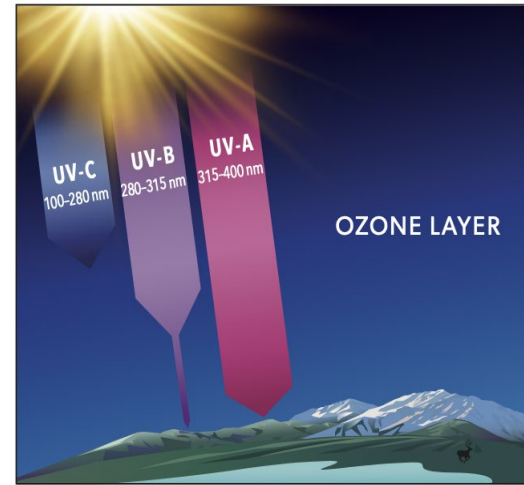
IPCC AR6 WG1 Technical Report, Figure TS15

- Chemistry and aerosols interact with the climate
- Importance of describing ozone and aerosol precursors
- Importance of aerosol-cloud interactions in models

Atmospheric Chemistry: Why is it important – Stratospheric Ozone



UV Protection by the Stratospheric Ozone Layer



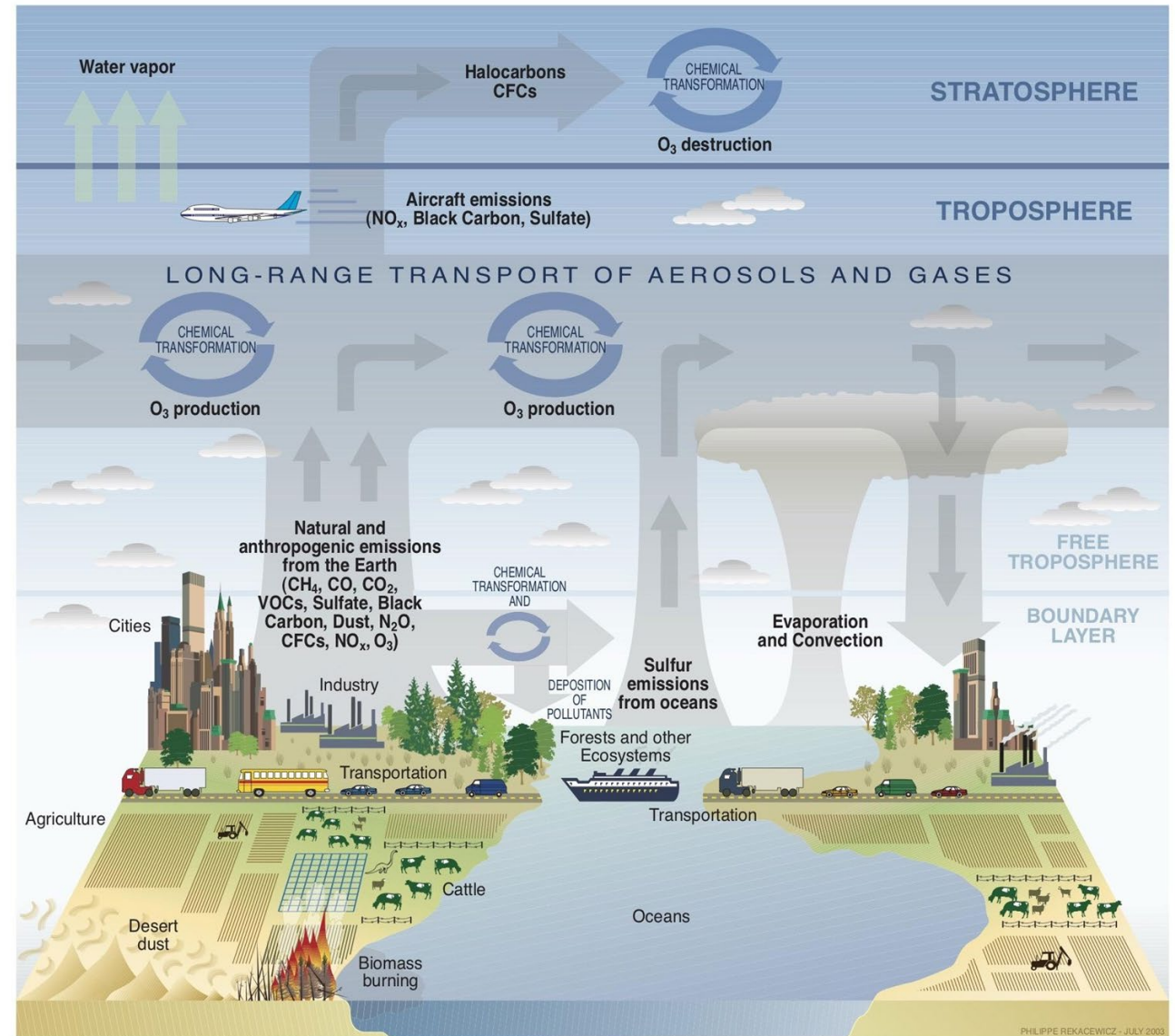
The ozone layer in the stratosphere protects life from harmful UV, through photochemical reactions

Accurate modeling is required:

- Impact on tropospheric chemistry
- Ozone hole recovery (CFCs)
- e.g. cause of a slowing trend

Atmospheric Chemistry

- Motivation
- Adding processes into models
 - Emissions
 - Chemical mechanism
 - Aerosol model and cloud interactions
 - Dry Deposition
 - Wet Deposition
- Applications
- Support



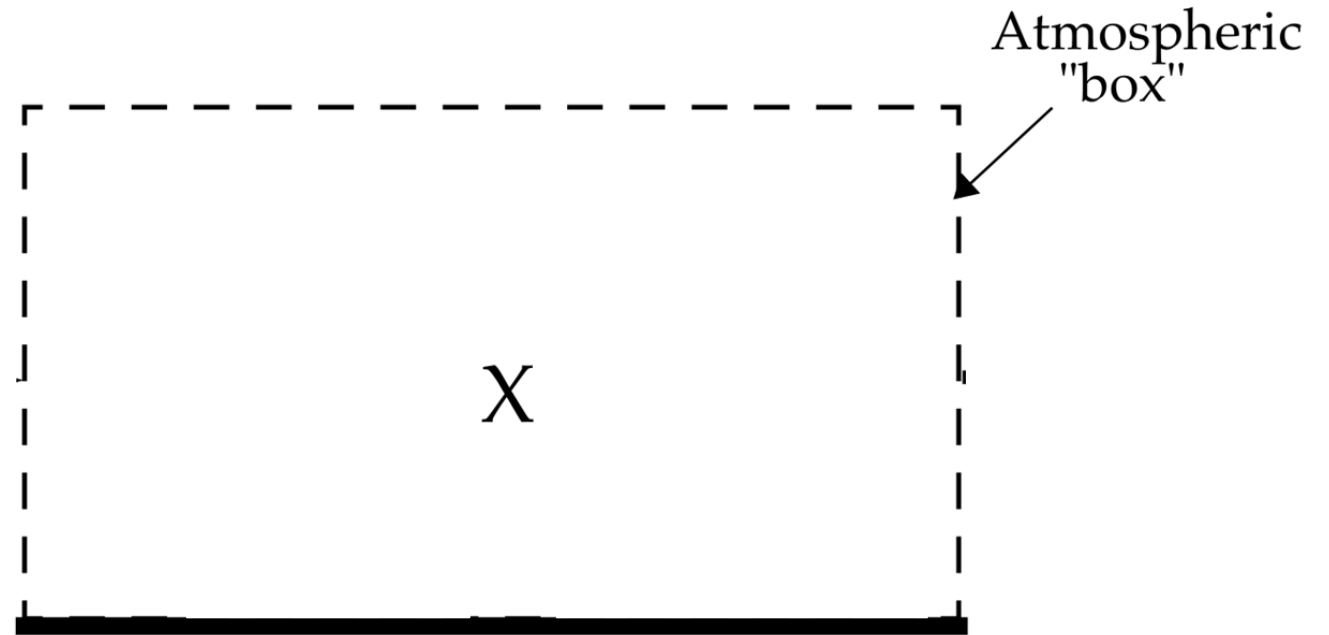
For each chemical constituent (χ), the following must be solved

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i)$$

Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

For each chemical constituent (χ), the following must be solved

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i)$$

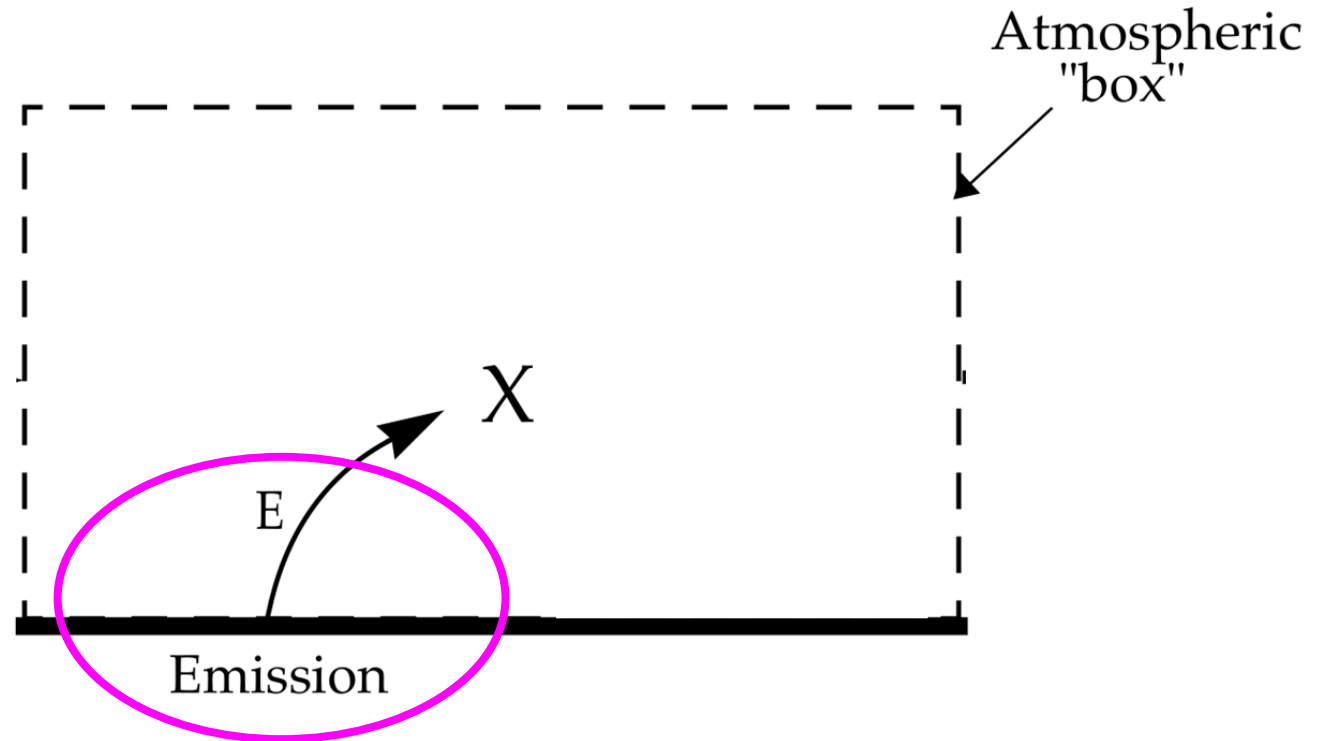


Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

For each chemical constituent (χ), the following must be solved

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i) = E_i + C_i + A_i + T_i - W_i - D_i$$

E_i Emissions



Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

Emissions in CESM: 4 main “types”

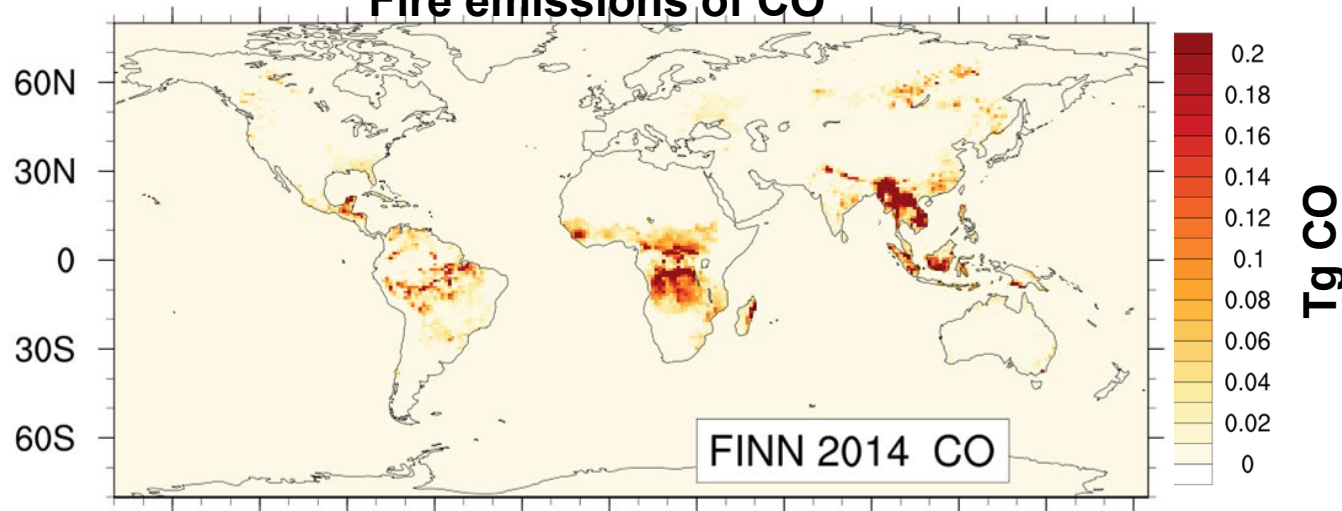
Emissions

- Surface emissions: anthropogenic, biogenic, biomass burning (fire), ocean, soil
- Vertical emissions: (external forcings): aircraft, volcanoes, power plants, (fire optional)
- Interactive: Dust, biogenic, sea salt, lightning NO_x , (fire optional/experimental)

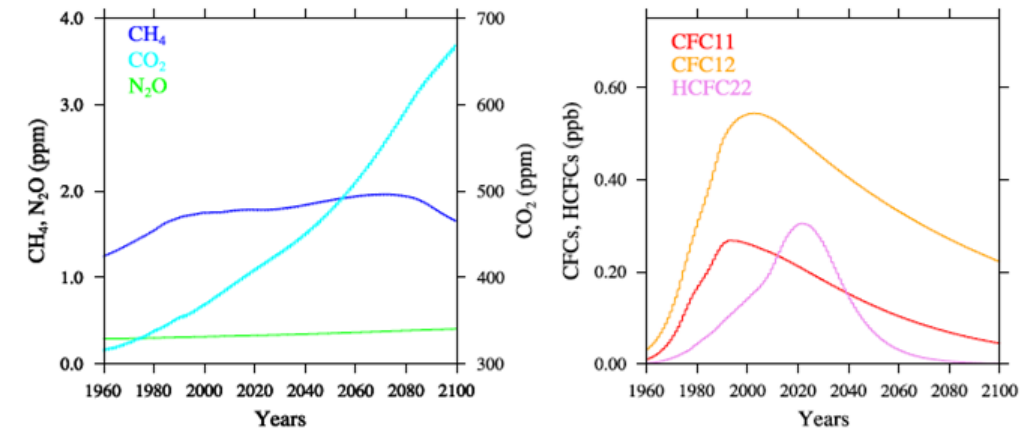
Surface concentrations

- Lower boundary conditions (greenhouse gases CO_2 , CH_4 , O_3 , N_2O and, long-lived gases CFCs). Can vary latitudinally.

Fire emissions of CO



Lower Boundary Conditions, RCP6.0



Interactive emissions: Dust

$$F_{dustemis} = F_{dustemis}(u_*, w)$$

(CESM default)

horizontal wind/friction

soil moisture

becomes

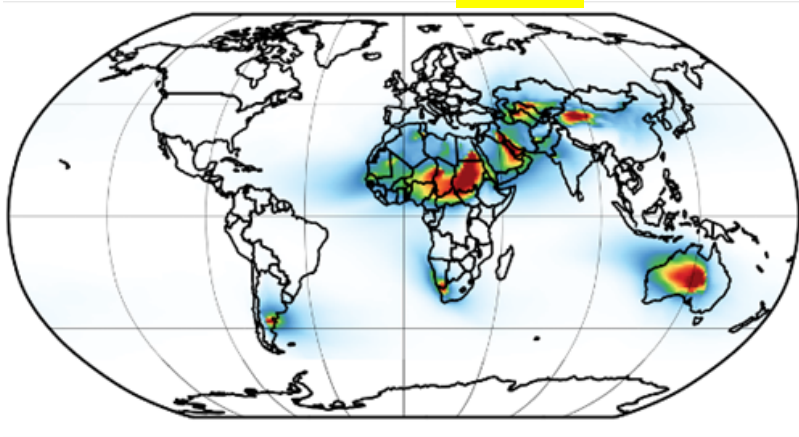
$$F_{dustemis} = F_{dustemis}(u_*, w, z_{0,rock}, LAI, \sigma_{\tilde{u}})$$

(Leung 2023)

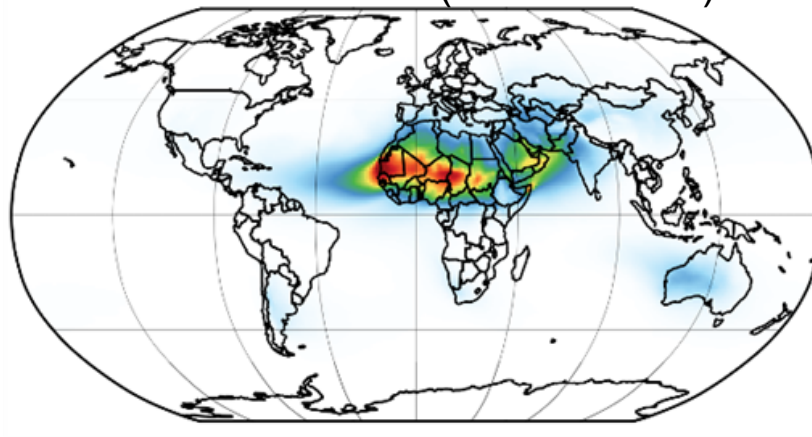
Drag partition due to surface roughness

Subtimestep wind following the similarity theory

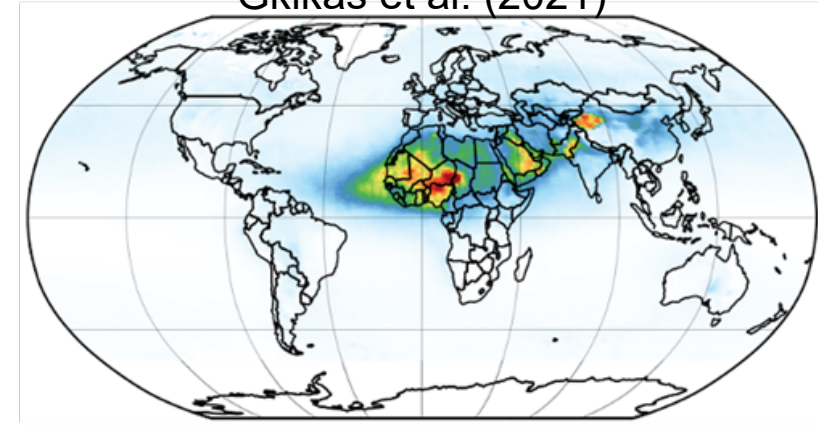
Charlie Zender et al. (2003; DEAD)
CESM2/CAM6 default



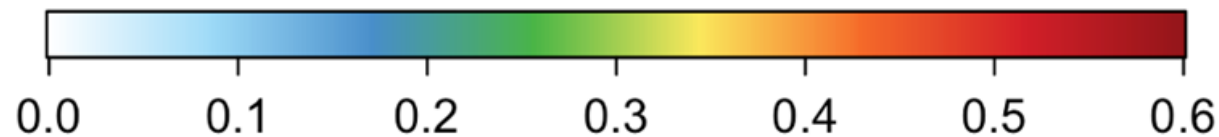
Danny Leung et al. (2023; L23)
CESM3/CAM7 (future default)



MIDAS (MODIS/Aqua) dust
Gkikas et al. (2021)



AOD from dust



Slide: Danny Leung

Interactive emissions: Biogenic

The **MEGAN-v2.1** algorithm

Emissions for species i :

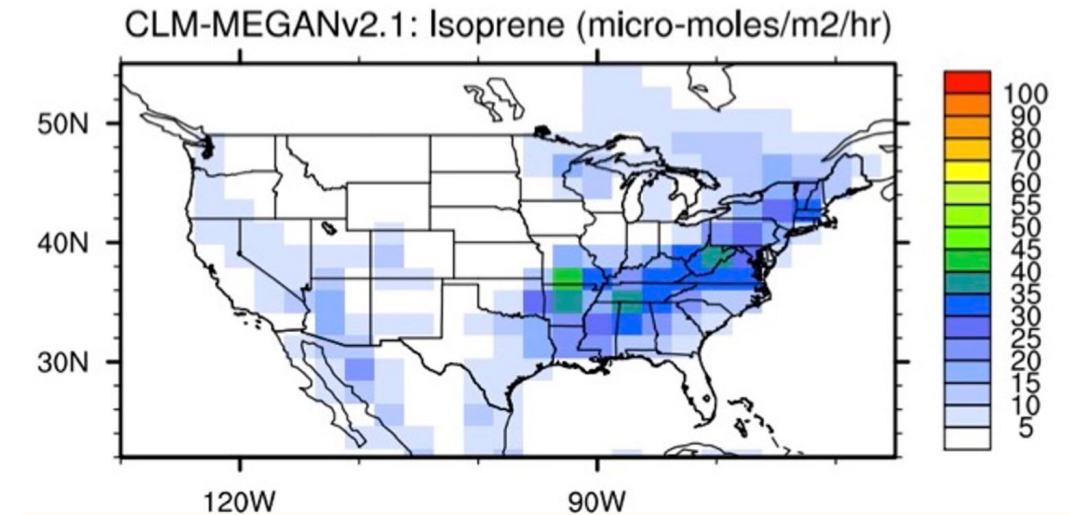
$$F_i = Y_i \sum_j \epsilon_{i,j} X_j$$

where

Y_i : emission activity factor, depends on leaf area index (LAI), meteorology (T, solar radiation), leaf age, soil moisture, with separate light-dependent and light-independent factors

$\epsilon_{i,j}$: emission factor at standard conditions for vegetation type (PFT) j

X_j : fractional area of PFT j



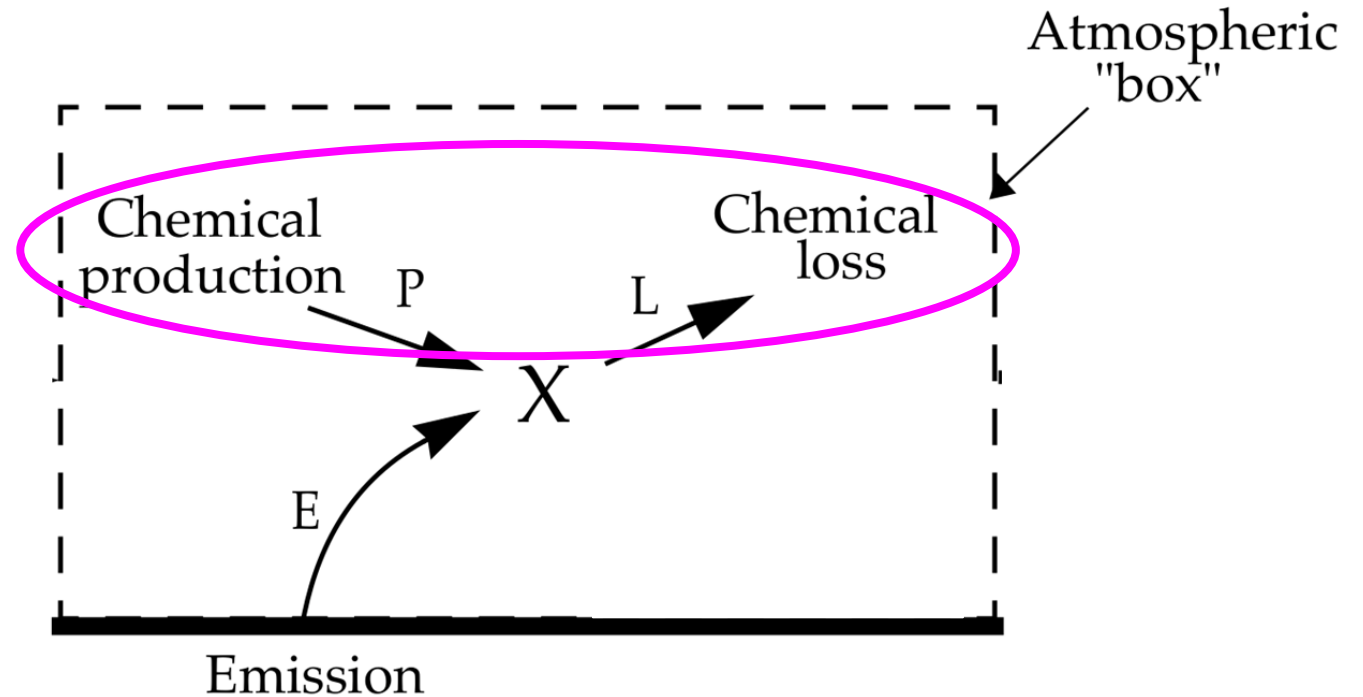
Guenther et al., GMD, 2012

Slide: Louisa Emmons

For each chemical constituent (χ), the following must be solved

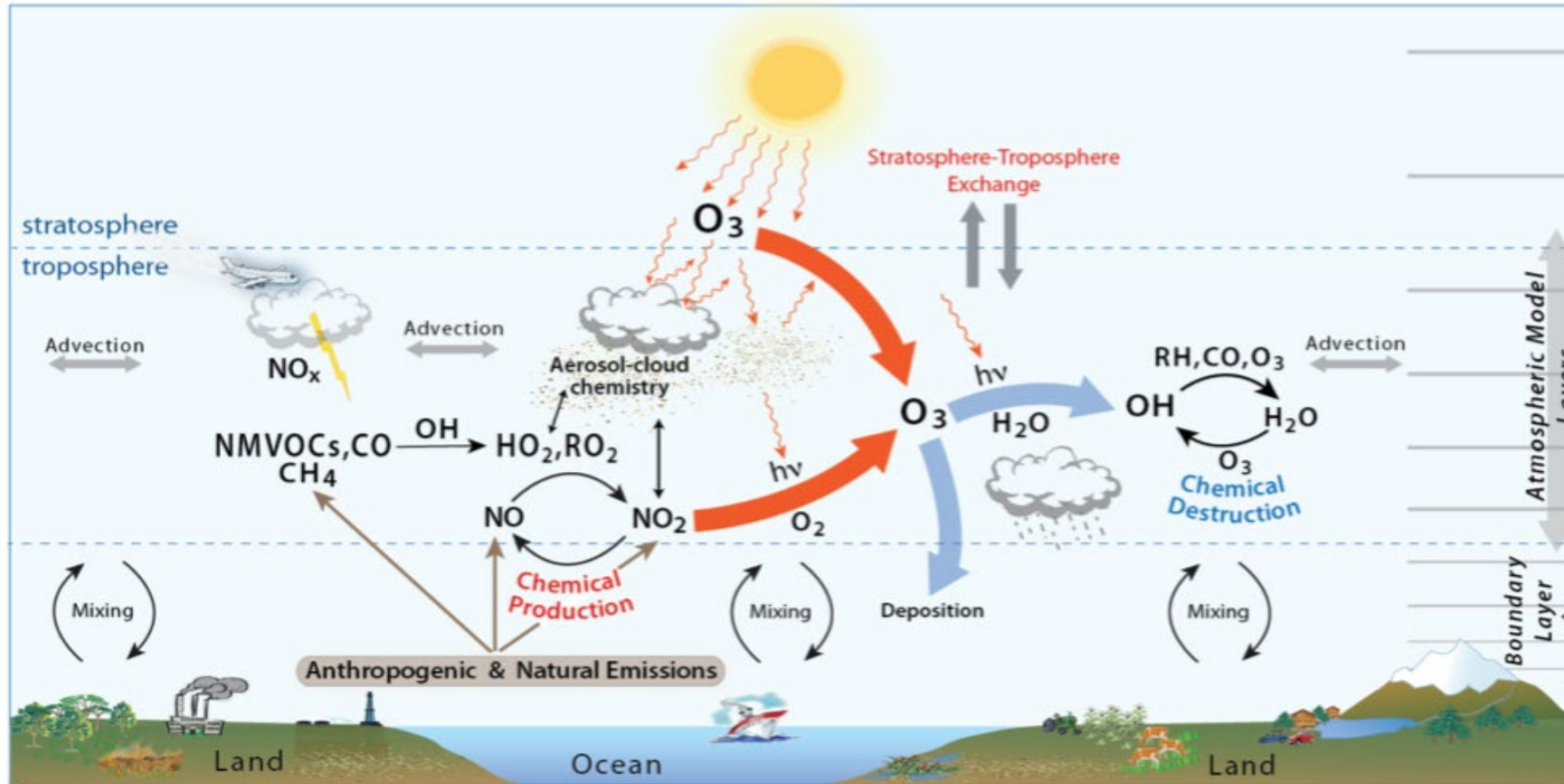
$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i) = E_i + C_i + A_i + T_i - W_i - D_i$$

E_i Emissions
 C_i Gas-phase-Chemistry
 A_i Aerosol-processes
(Gas-aerosol exchange,
het chem.)



Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

Tropospheric Chemistry



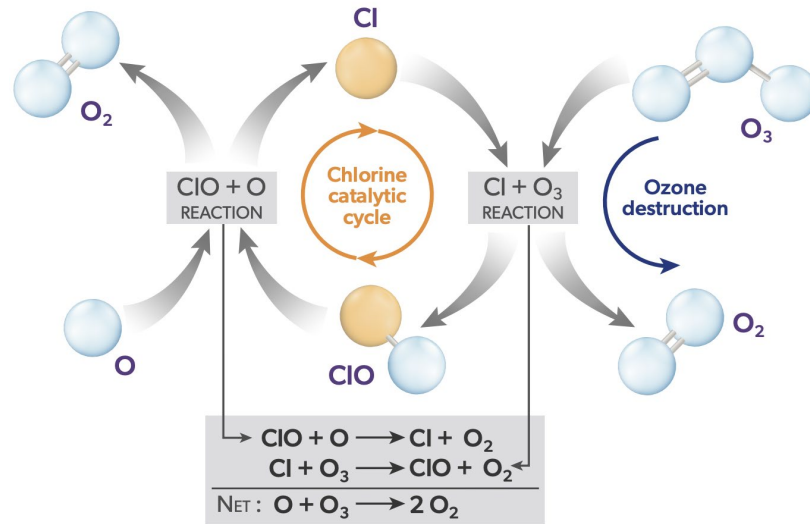
Photochemistry
Gas-phase chemistry
Heterogeneous chemistry
Aqueous phase chemistry
Gas-to-aerosol Exchange

Young et al., 2018, <https://doi.org/10.1525/elementa.265>

Stratospheric Chemistry

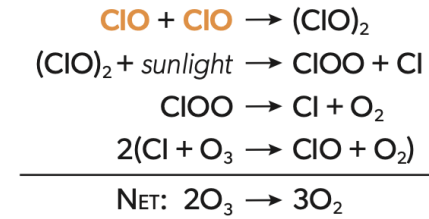
WMO2022

Ozone Destruction Cycle 1 : Upper Stratosphere

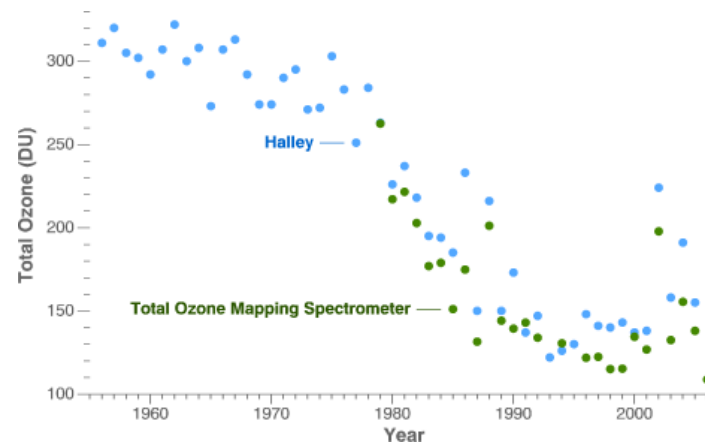
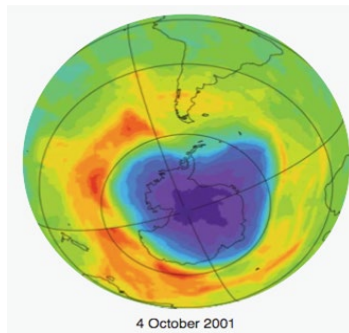
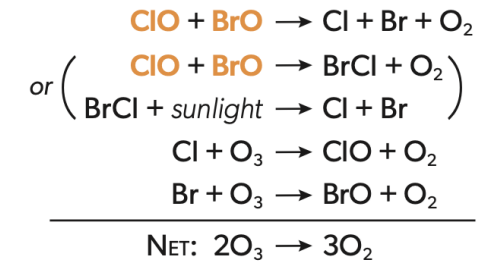


Ozone Destruction Cycles 2 and 3 : Polar Regions

Cycle 2 :



Cycle 3 :

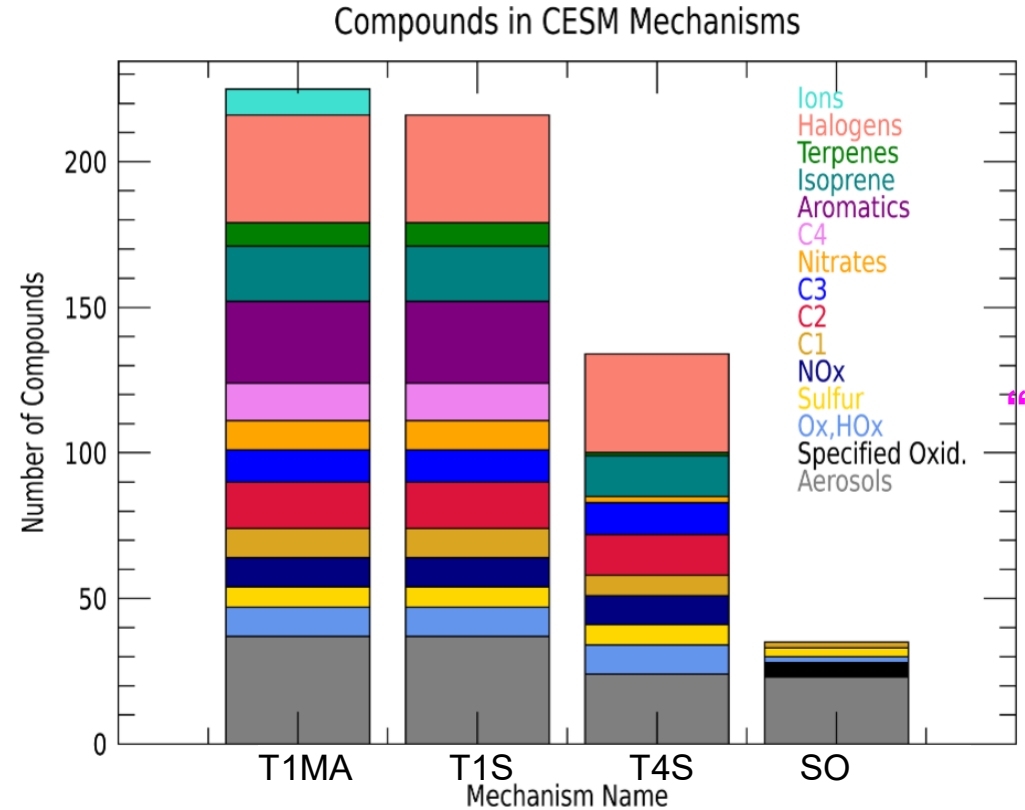


- Comprehensive Stratospheric Chemistry
- Heterogeneous reactions
 - Catalytic Cycles

Atmospheric chemistry mechanisms in CESM

Chemistry mechanism descriptions:

<https://www2.acom.ucar.edu/gcm/mozart>



T1S = default
“full-chemistry”
Troposphere
and
Stratosphere

Name	Description	# species
T1	Comprehensive tropospheric chemistry; for air quality simulations	179
T2	T1 with detailed terpene chemistry	265
T4	Simpler tropospheric chemistry suitable for climate simulations	97
T1S	T1 with comprehensive stratospheric chemistry	216
T1MA (TSMLT)	T1 with stratosphere, mesosphere, lower thermosphere chemistry	225
T4S	T4 with comprehensive stratospheric chemistry	134
SO	Specified Oxidants, with GHGs	33

Slide: Louisa Emmons

CAM6 (Specified Oxidants, SO) vs CAM -chem

Same atmosphere, physics, resolution

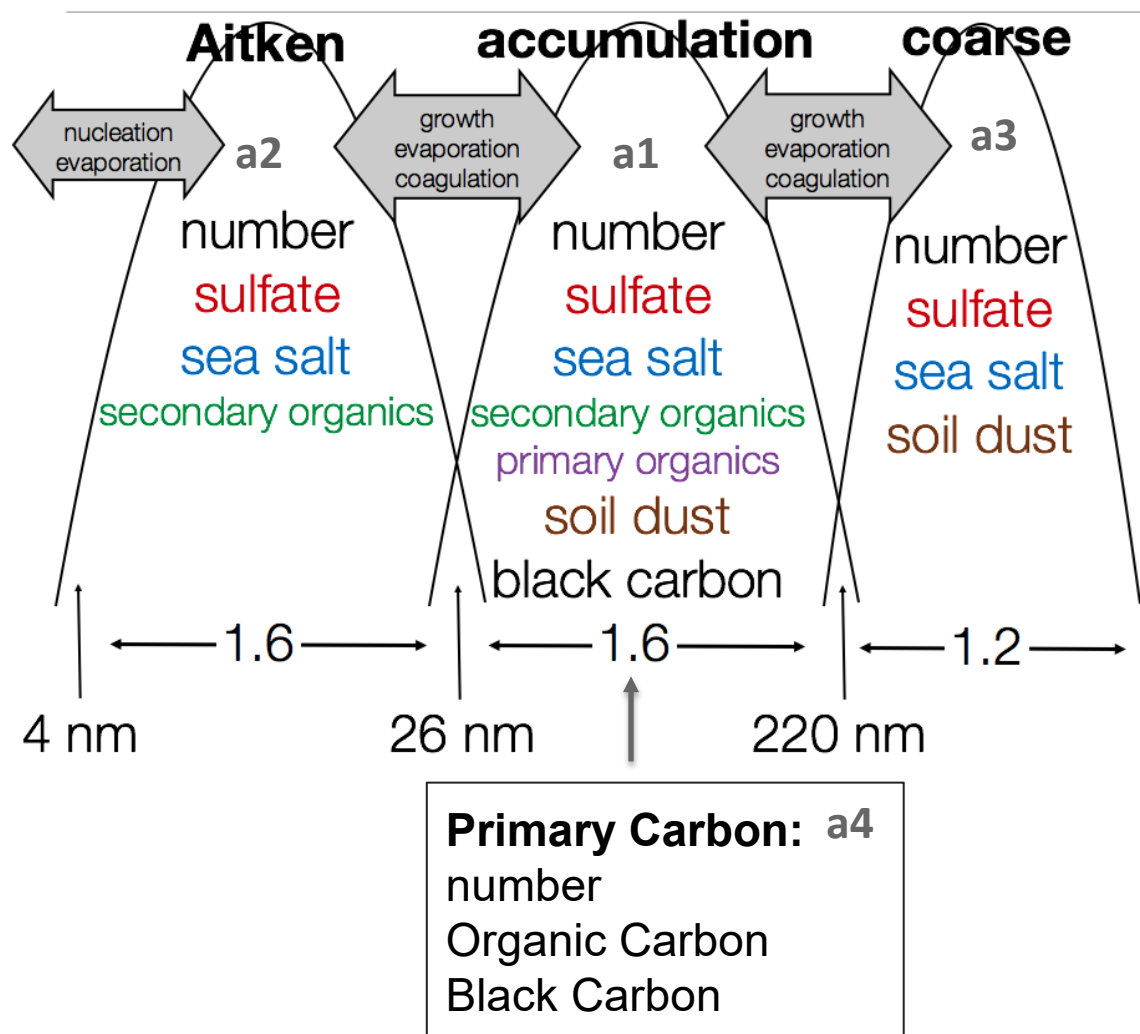
Different chemistry and aerosols -> emissions and coupling

- **CAM6:** Aerosols are calculated, using simple chemistry (with “fixed” oxidants) (prescribed: N_2 , O_2 , H_2O , O_3 , OH , NO_3 , HO_2 ; chemically active: H_2O_2 , H_2SO_4 , SO_2 , DMS, SOAG)

Limited interactions between Chemistry and Climate

- > prescribed fields are derived using chemistry -climate simulations
- Prescribed ozone is used for radiative calculations
- Prescribed oxidants is used for aerosol formation
- Prescribed methane oxidation rates
- Prescribed stratospheric aerosols
- Prescribed nitrogen deposition
- Simplified secondary organic aerosol description

Default Modal Aerosol Model (MAM4)



Representation of

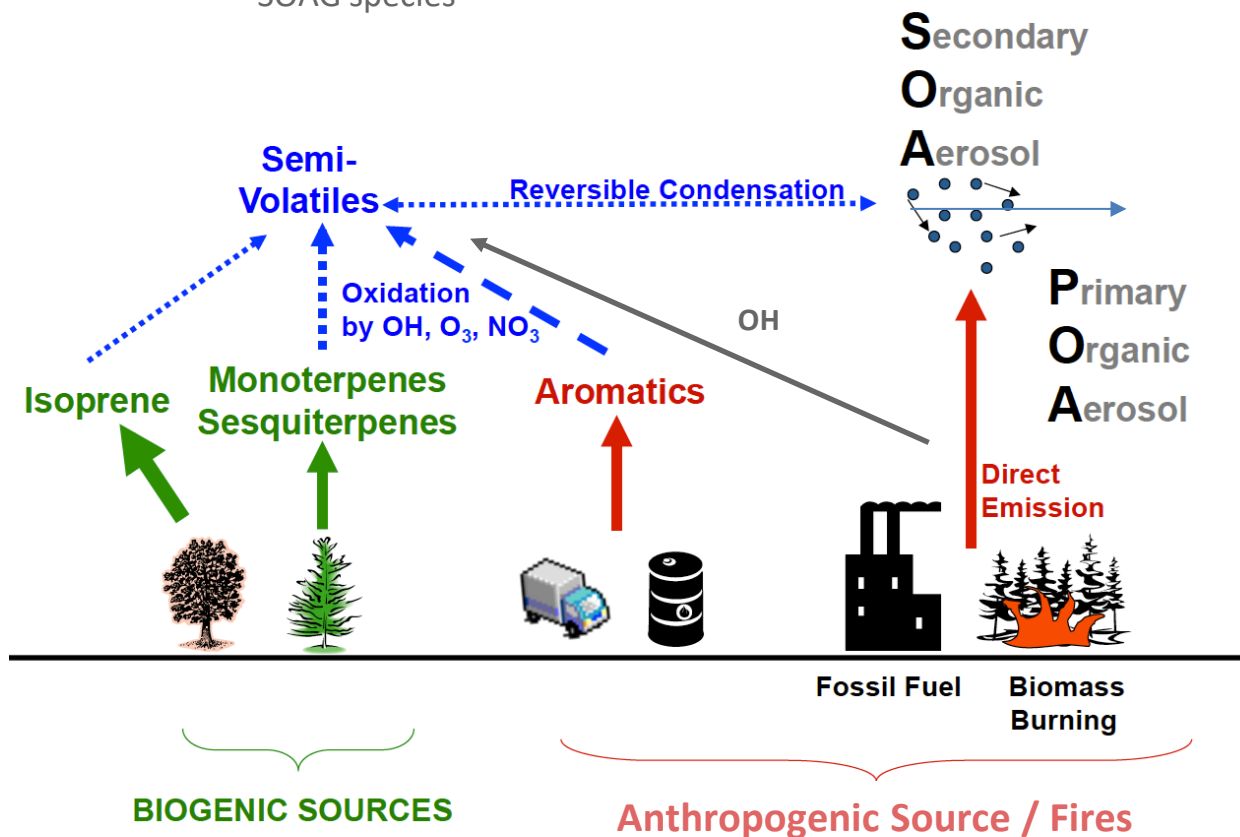
- Sulfates,
- Black Carbon
- Organic Carbon, Organic Matter (OC, SOA),
- Mineral Dust and Sea-Salt

Slide: Mike Mills

Secondary Organic Aerosol Description

ORGANIC CARBON AEROSOL SOURCES

SOAG species



Simplified Chemistry (CAM6):

- SOAG (oxygenated VOCs) derived from fixed mass yields
- no interactions with land

Comprehensive Chemistry:

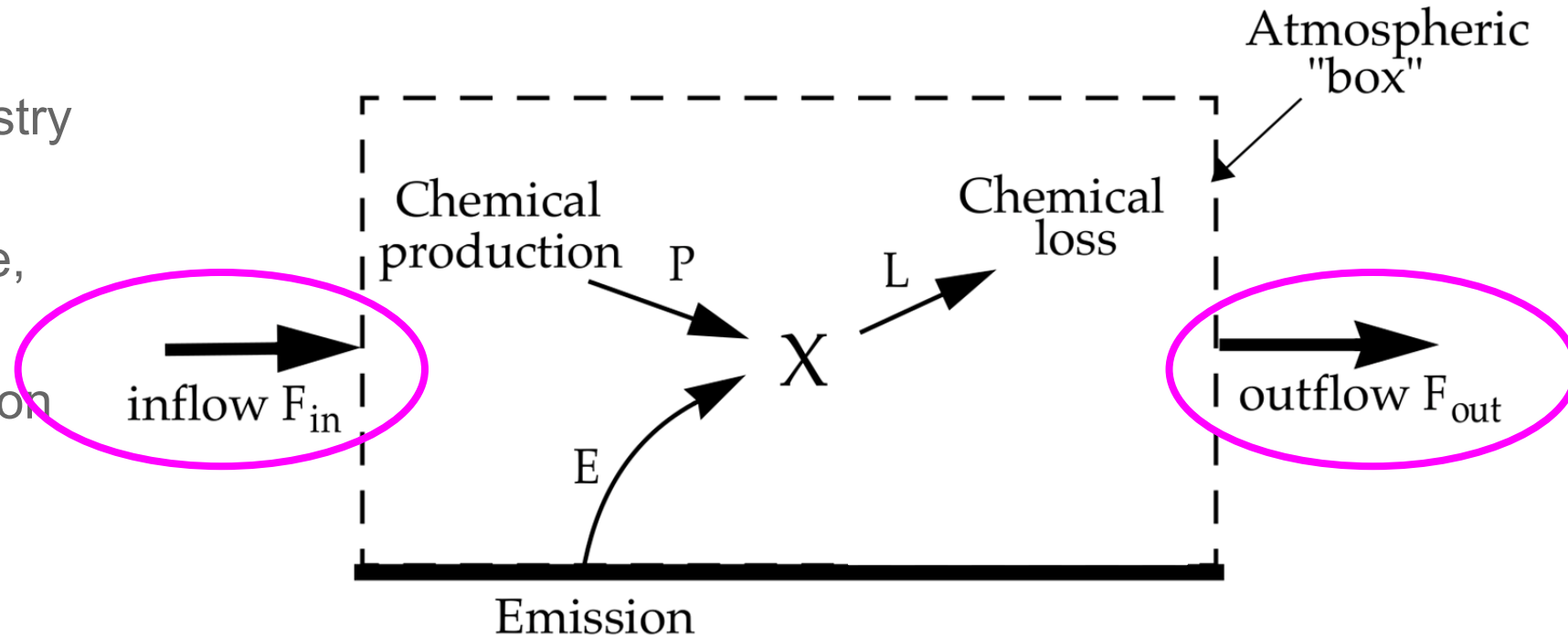
- SOAG formation derived from VOCs using Volatility Bin Set (VBS)
 - 5 volatility bins
 - Interactive with land emissions
- > a more physical approach

Modified from C. Heald, MIT Cambridge

For each chemical constituent (χ), the following must be solved

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i) = E_i + C_i + A_i + T_i - W_i - D_i$$

E_i Emissions
 C_i Gas-phase-Chemistry
 A_i Aerosol-processes
(Gas-aerosol exchange,
het chem.)
 T_i Advection + Diffusion



Free running versus nudged (T, U, V)

Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

Dynamical core reminder

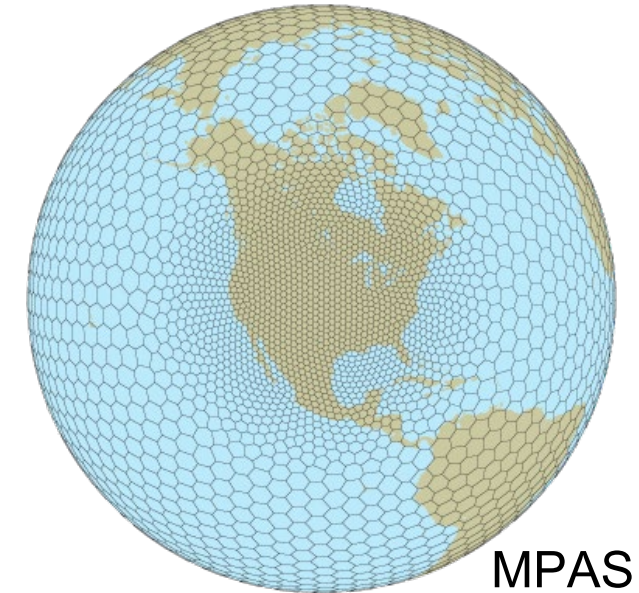
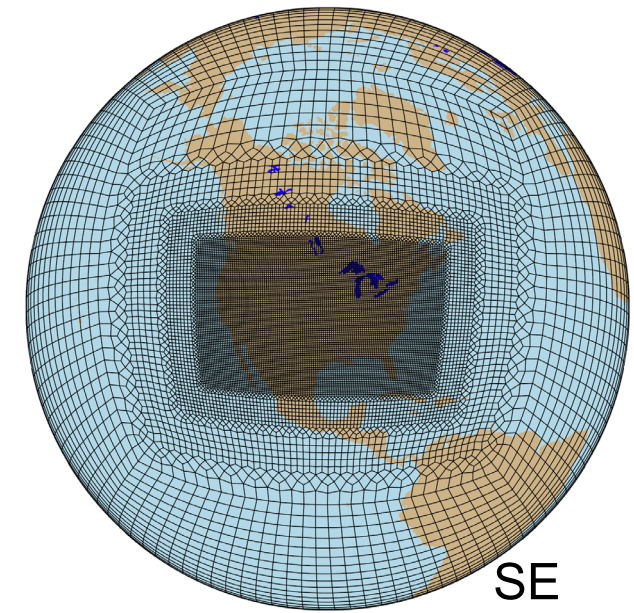
FV: Finite Volume (FV) “regular grid”

FV3: a non-hydrostatic cubed -sphere version of FV

SE - CSLAM (pg3): Spectral Element dynamical core on a cubed sphere, Conservative Semi-Lagrangian Multi-tracer dynamical core with finite-volume transport (CSLAM). No current regional refined capability.

SE (RR): Spectral Element dynamical core with regional refinement options.

MPAS: Model for Prediction Across Scales, cloud resolving, a global version of Weather Research and Forecasting, WRF, model discretized on a Voronoi grid. Regional refinement option, (experimental in CESM: need to compare with SE-RR).

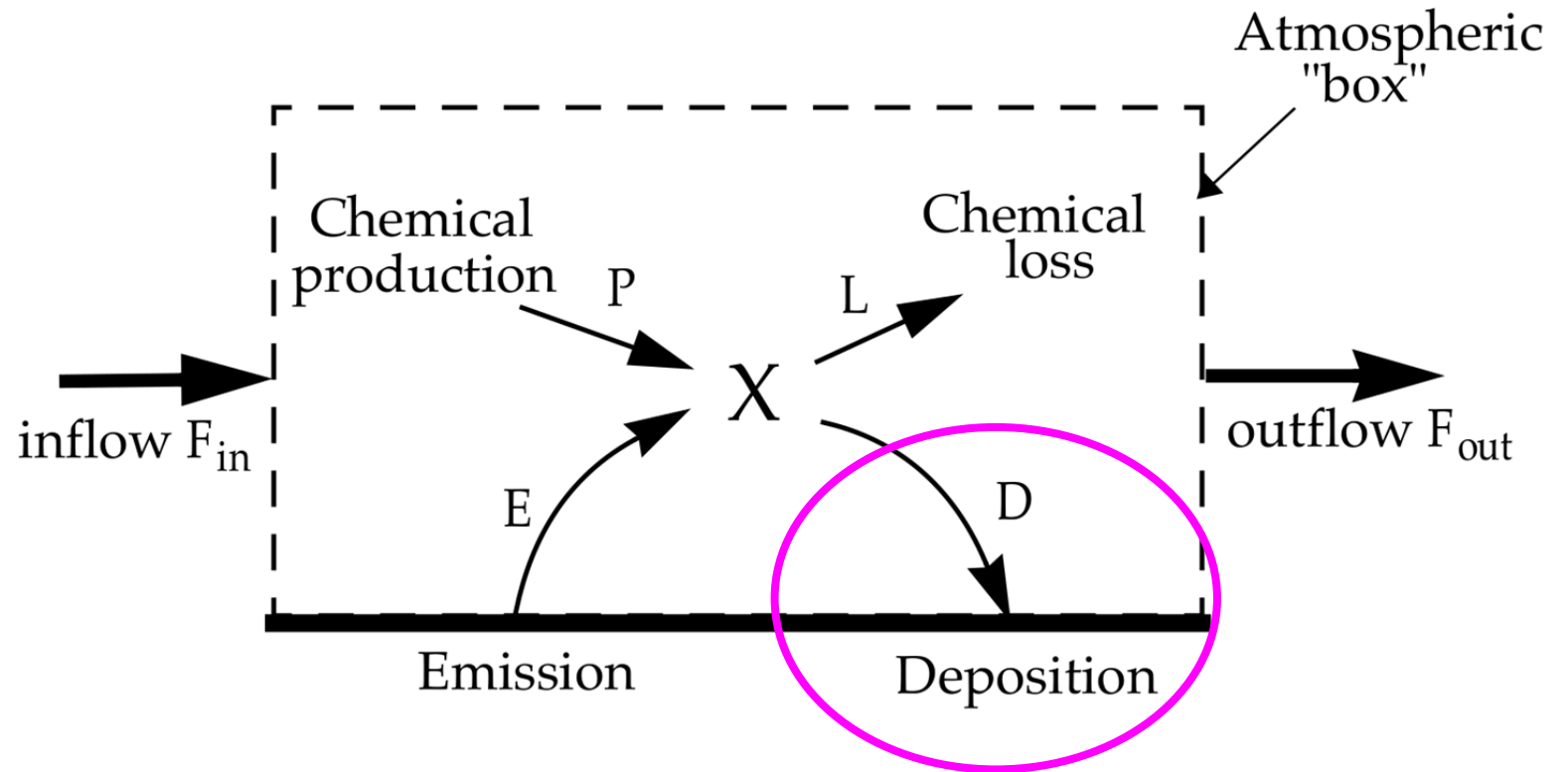


<https://www.cesm.ucar.edu/sites/default/files/2024-08/2024cesmtutorialauritzen.pdf>

For each chemical constituent (χ), the following must be solved

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i) = E_i + C_i + A_i + T_i - W_i - D_i$$

- E_i Emissions
 C_i Gas-phase-Chemistry
 A_i Aerosol-processes
(Gas-aerosol exchange,
het chem.)
 T_i Advection + Diffusion
 W_i Cloud-processes
(wet deposition)
 D_i Dry deposition



Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

Wet Deposition

Large-scale and convective precipitation: uptake of chemical constituents in rain or ice

Considers in-cloud and below-cloud scavenging rates and solubility factors of aerosol and chemical species

A first-order loss process

$$\chi_{iscav} = \chi_i \times F \times (1 - \exp(-\lambda \Delta t))$$

χ_{iscav} scavenged species (kg)

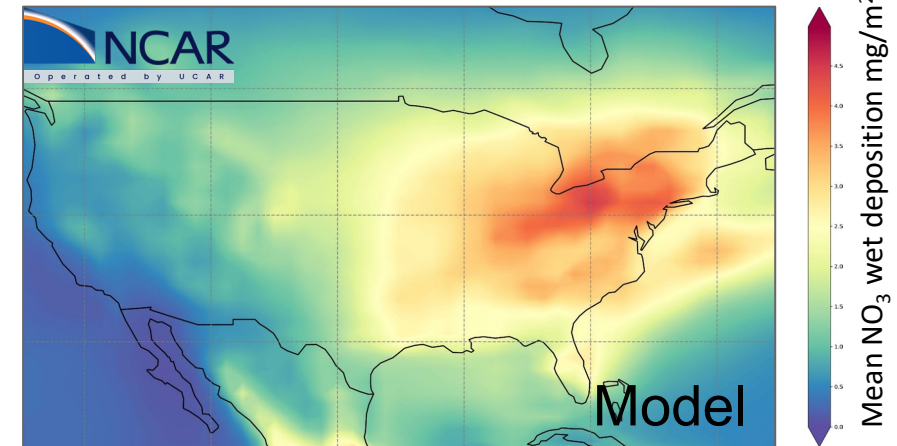
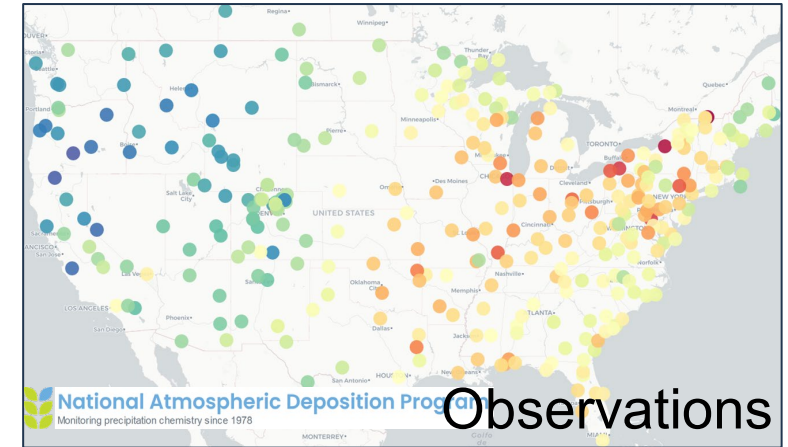
χ_i species

F fraction of the grid box from which tracer is being removed

λ is the loss rate

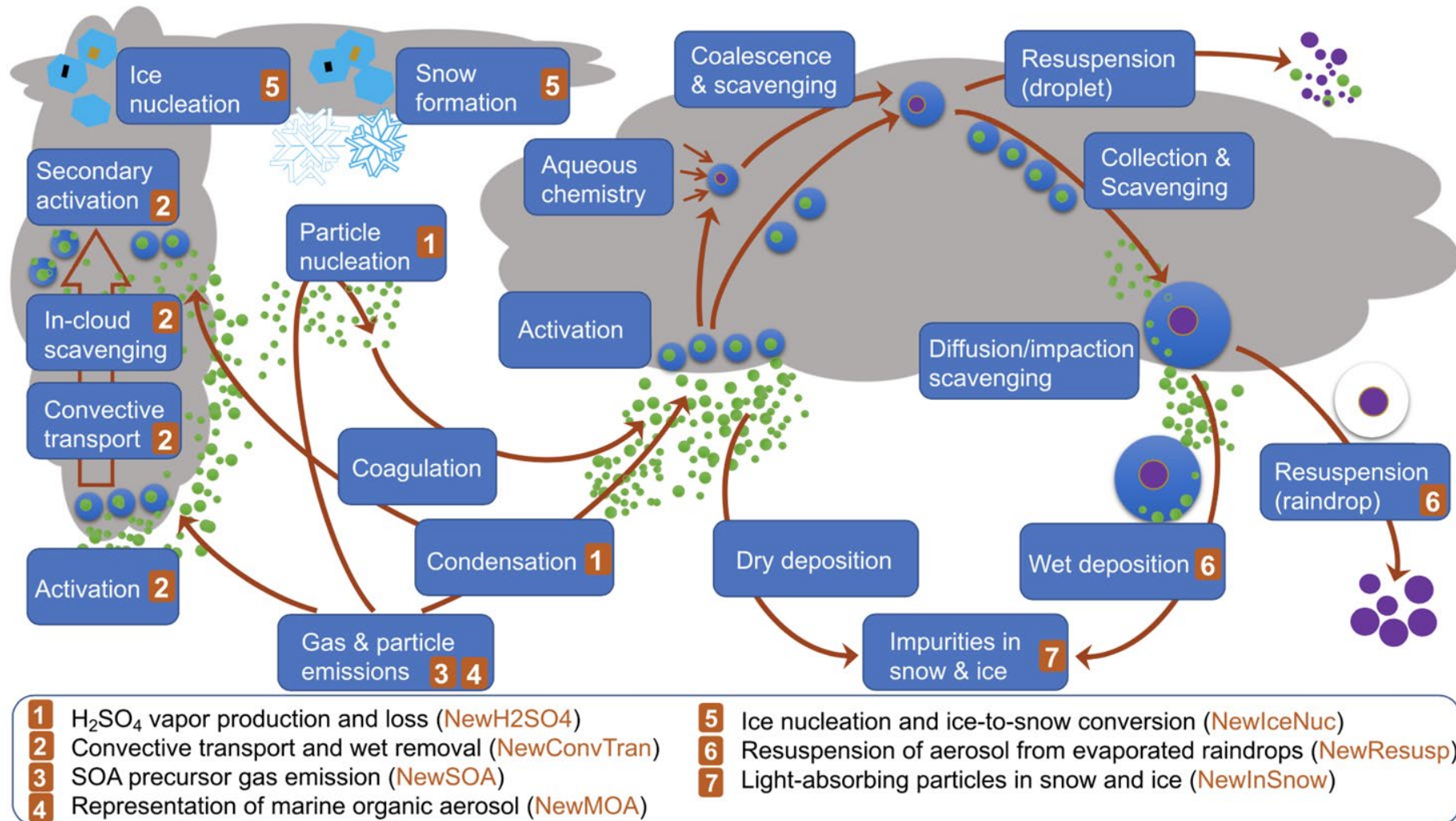


Deni
Murray
ACOM
ASP
graduate
visitor



References: (Barth et al., 2000, Neu and Prather 2012, Lamarque et al., 2012)

Aerosol – Cloud Interactions



E3SM: Wang et al., 2020 (JAMES)

Dry Deposition Velocity Calculation

Resistance model:

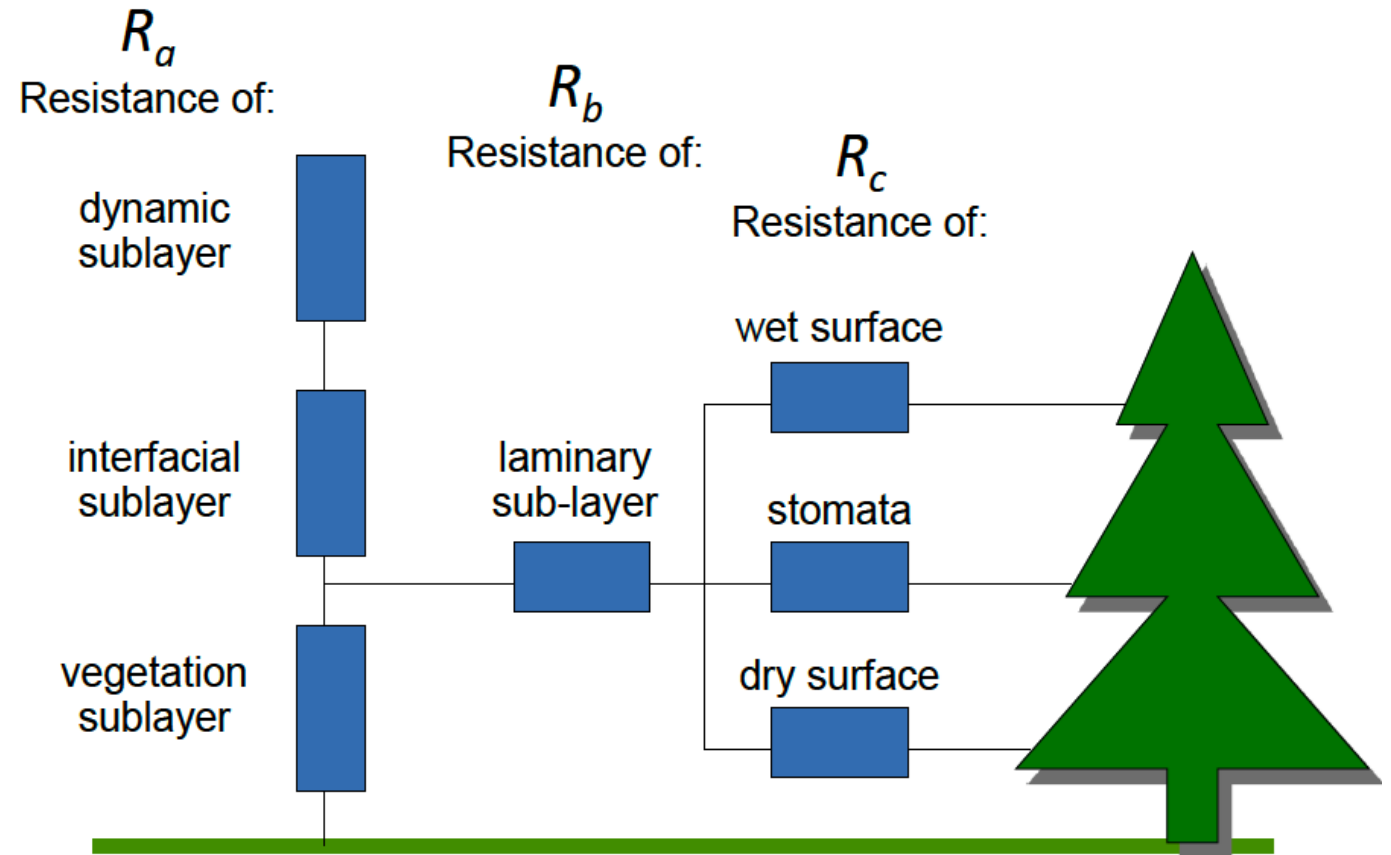
$$V_d = \frac{1}{R_a + R_b + R_c}$$

$$F = -v_d C$$

F = deposition flux

C = concentration of species in
10m surface layer

Uptake of chemical constituents by
plants and soil (CLM), depends on
land type, roughness of surface

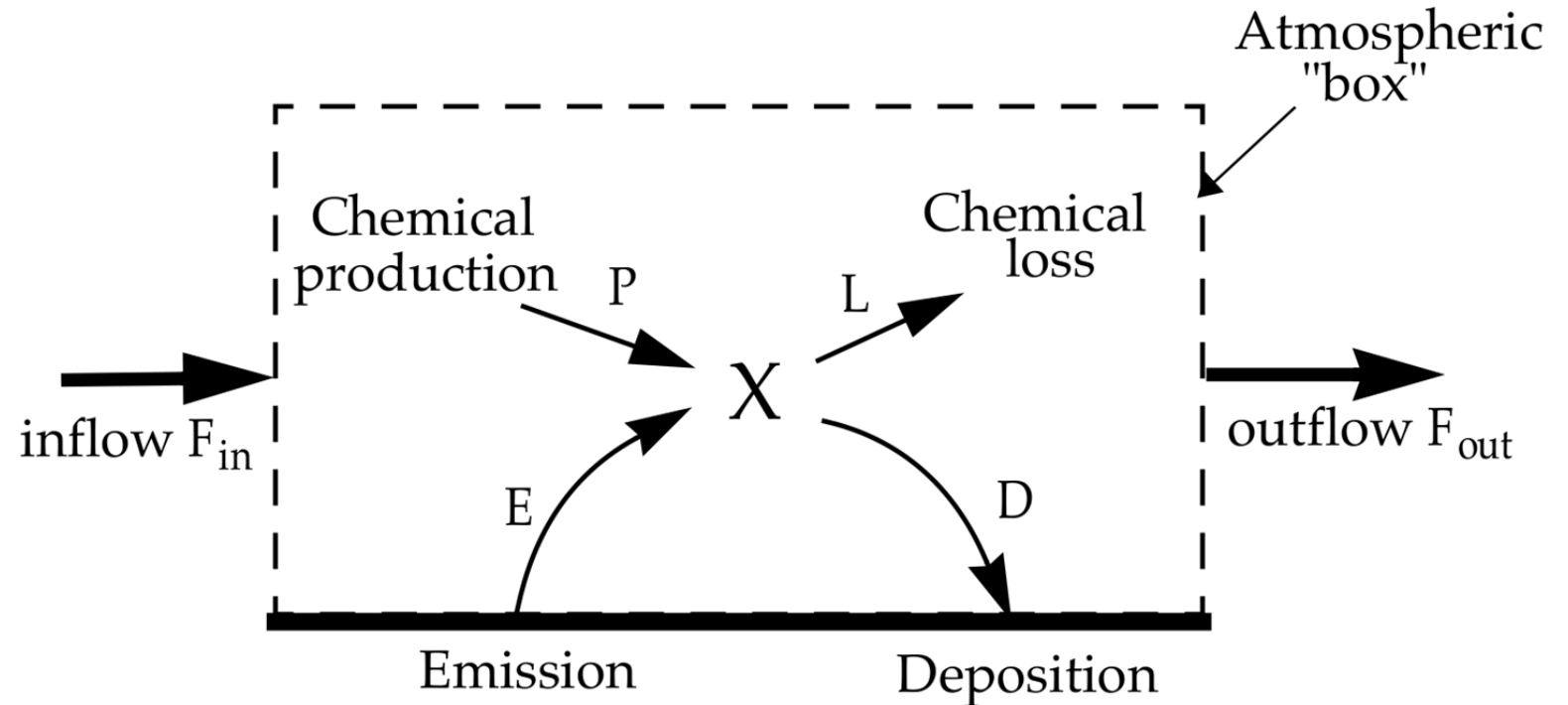


For each chemical constituent (χ), the following must be solved

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i) = E_i + C_i + A_i + T_i - W_i - D_i$$

it can get expensive very fast! \$\$\$

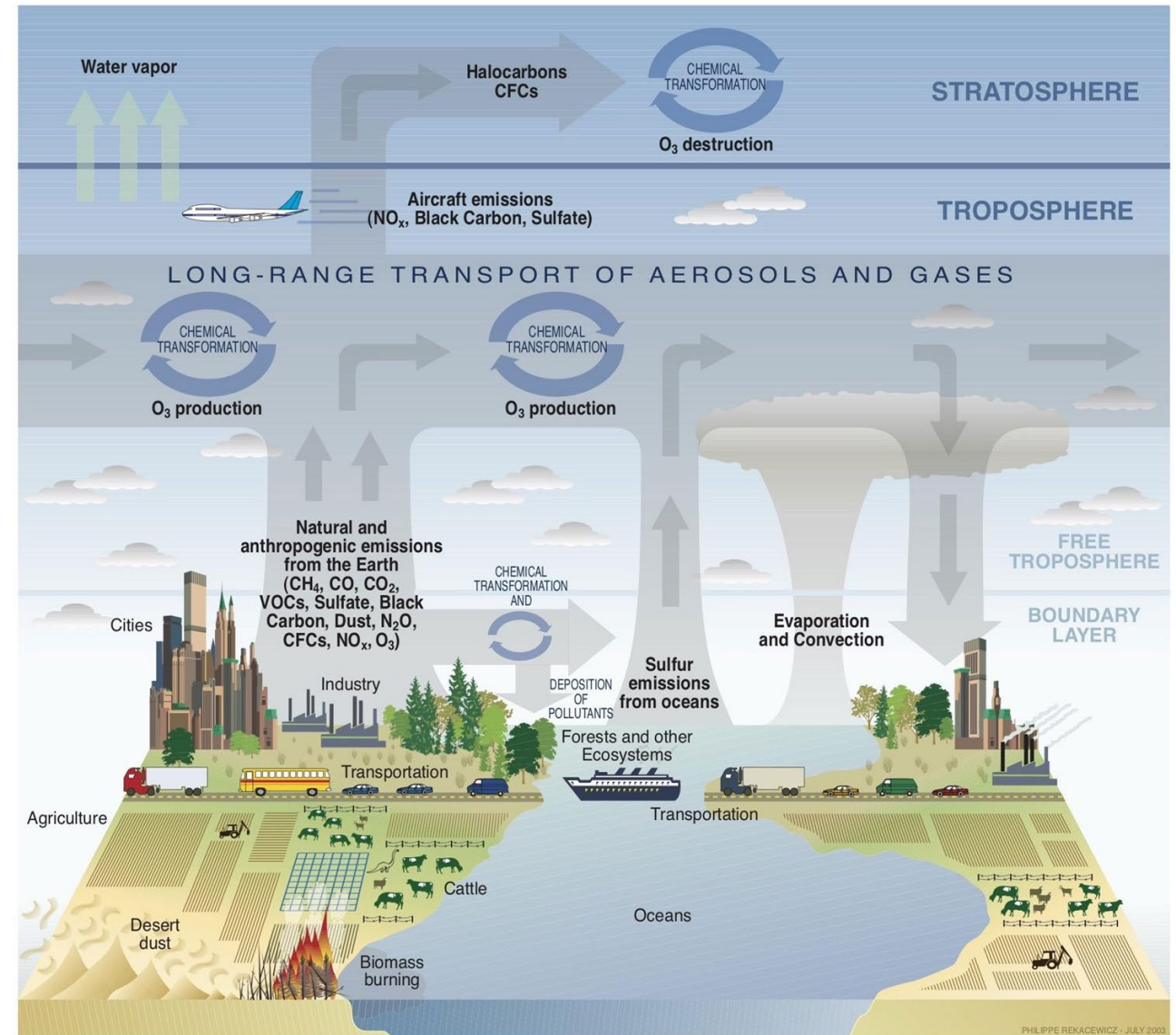
- E_i Emissions
- C_i Gas-phase-Chemistry
- A_i Aerosol-processes
(Gas-aerosol exchange,
het chem.)
- T_i Advection + Diffusion
- W_i Cloud-processes
(wet deposition)
- D_i Dry deposition



Introduction to Atmospheric Chemistry, Daniel J. Jacob <https://acmg.seas.harvard.edu/education/introduction-atmospheric-chemistry>

Atmospheric Chemistry

- Motivation
- Adding processes into models
 - Emissions
 - Chemical mechanism
 - Aerosol model and cloud interactions
 - Dry Deposition
 - Wet Deposition
- Applications: CAM-chem
- Support



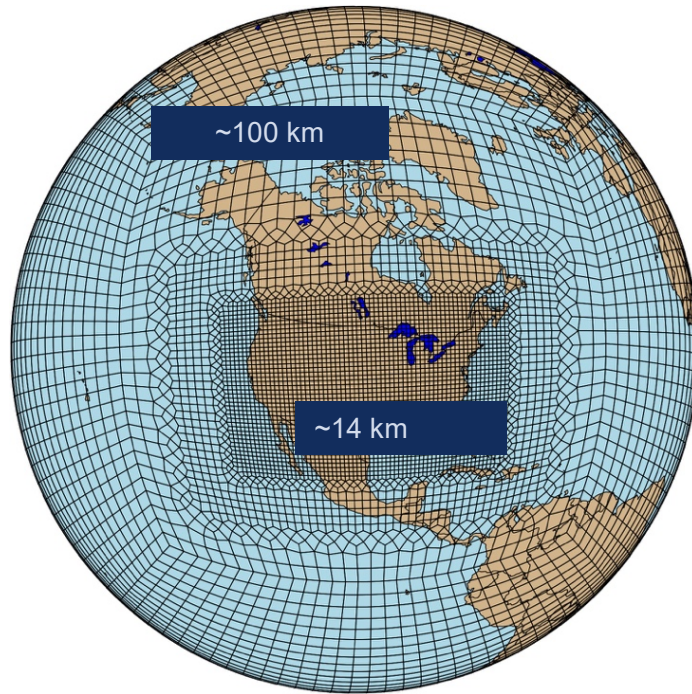
Chemistry → Air Quality: Regional refinement

MUSICA-V0: Multi -Scale Infrastructure for Chemistry and Aerosols

CAM-chem-SE-RR - Community Atmosphere Model with Chemistry With Spectral Element (SE) dynamical core and Regional Refinement (RR)

MUSICA-wiki: tutorials and support

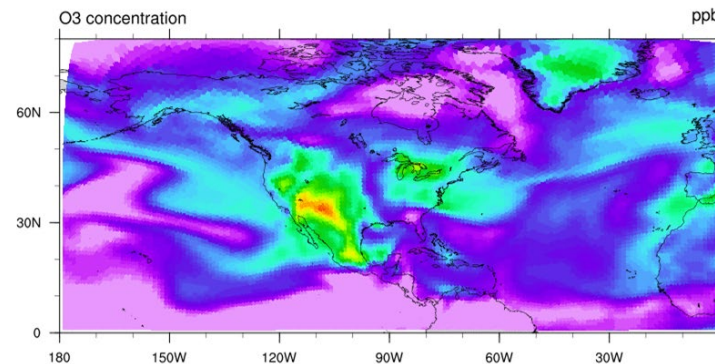
<https://wiki.ucar.edu/display/MUSICA>



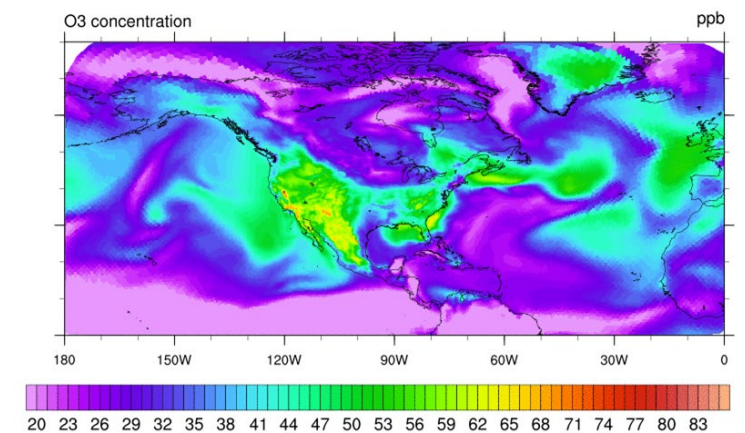
Example: U.S. Air Quality, Surface Ozone (ppb)

- Exposure Relevant scales and large-scale feedbacks

Global 1 degree



Regional Refined

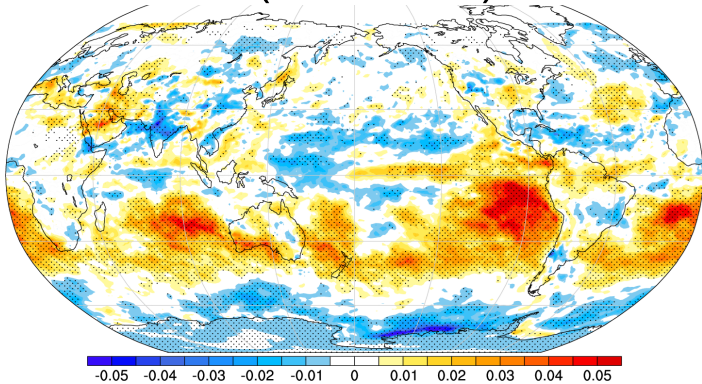


Chemistry → Climate: Australian wildfires 2019/2020

- CESM/CAM6 simulation with aerosols, satellite-based inventory (GFED) in Australia compared to climatology
- Climate response similar to a major volcanic eruption (aerosol-cloud interactions)
- Large interhemispheric radiative imbalance anomaly and impacts on ENSO

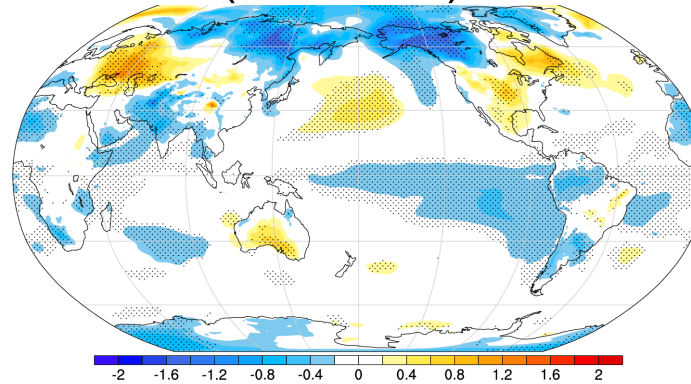


Cloudy Sky Albedo
(Jan 2020)

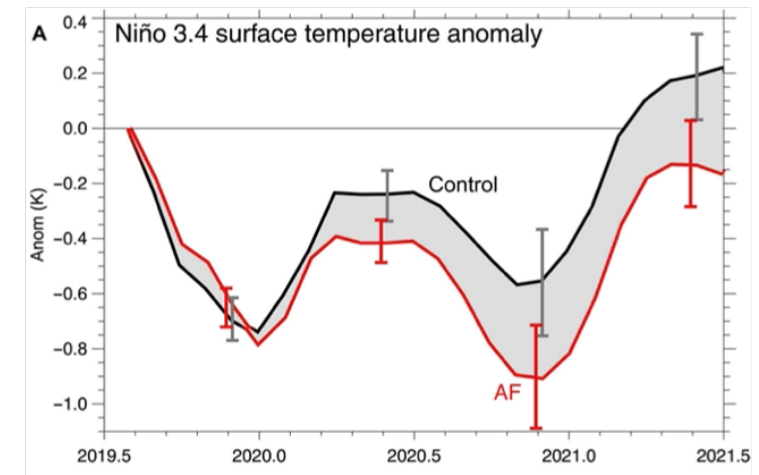


Cloud brightening across
the Southern Hemisphere

Near-surface temperature
(Jan 2021)



2020/21 La Niña response

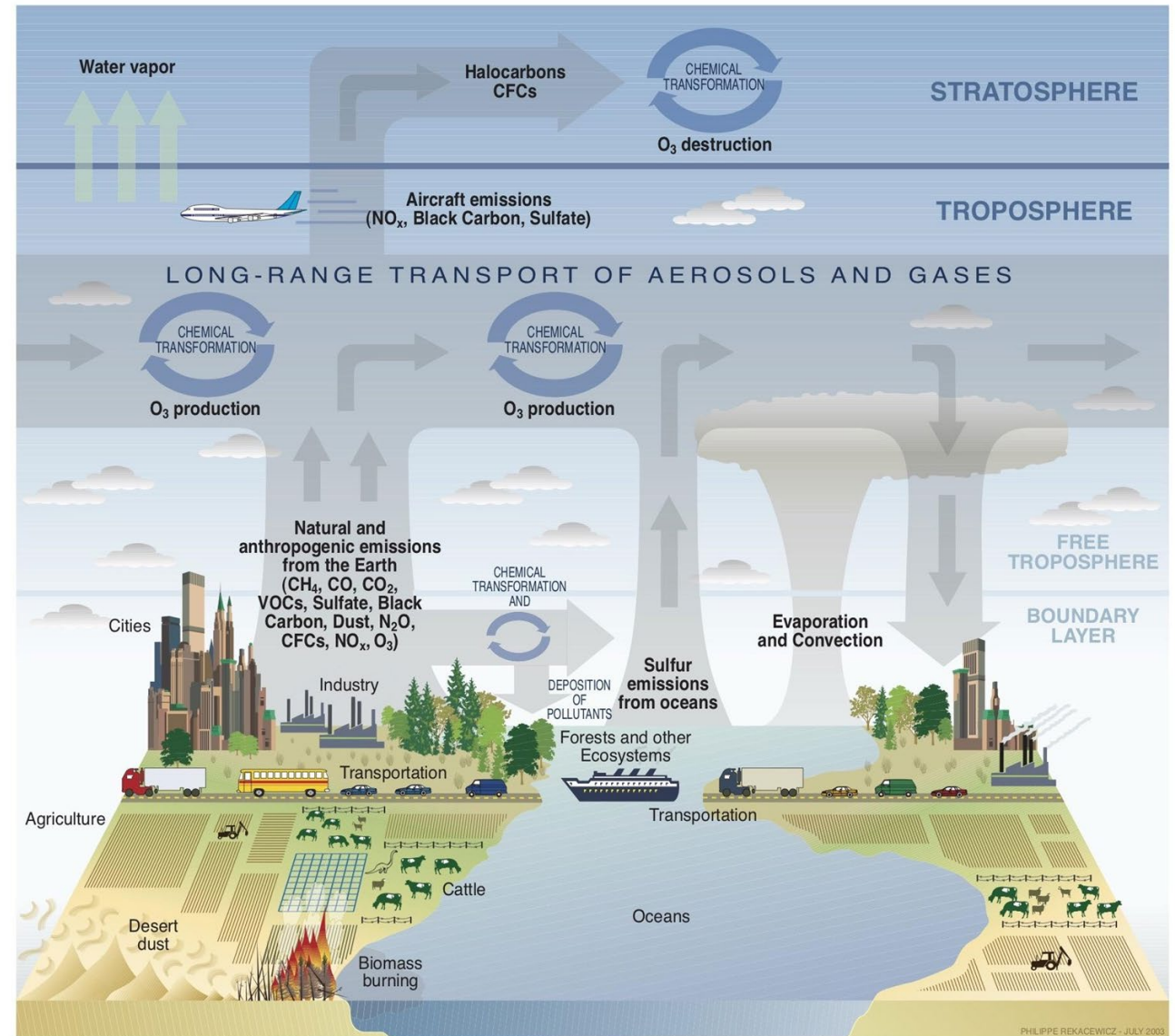


Fasullo et al., GRL, 2021

Fasullo et al., Sci. Adv., 2023

Atmospheric Chemistry

- Motivation
- Adding processes into models
 - Emissions
 - Chemical mechanism
 - Aerosol model and cloud interactions
 - Dry Deposition
 - Wet Deposition
- Applications
- Support



User Support for chemistry modeling in CESM

Wiki Page for Chemistry:

<https://wiki.ucar.edu/display/camchem/Home>

Use and Diagnostics	<ul style="list-style-type: none">• Boundary conditions for regional modeling• Atmospheric Diagnostics (ADF) in python NEW!• Automated CESM diagnostic package (using NCL)• Using CAM-chem output• MELODIES MONET model-obs comparison package
User Community	<ul style="list-style-type: none">• Current Users/Projects• Contributions to Model Intercomparisons (MIPs)• CAM-chem Forum• Chemistry-Climate Working Group Publications• CAM-chem Publications from NCAR• CESM Publications
Other links and documents	<ul style="list-style-type: none">• Recent Bug Fixes• CAM Documentation (User and Scientific Guides)• ACOM CAM-chem page• CESM Chemistry Climate Working Group• Join the CESM Chemistry WG mailing list• Benchmarks and Production Experiment Diagnostics

Forum to search for and ask questions:

[http://bb.cgd.ucar.edu /](http://bb.cgd.ucar.edu/)

Contact us:

Simone Tilmes

CAM-Chem co-chair

tilmes@ucar.edu

Rebecca Buchholz

CAM-Chem Liaison

buchholz@ucar.edu

Shawn Honomichl

CAM-Chem Liaison

shawnh@ucar.edu

Regional Refinement Wiki: <https://wiki.ucar.edu/display/MUSICA>



Key takeaways

- Atmospheric chemistry is important in models due to the feedback into the earth system. It has **impacts on health, weather and climate** .
- Adding atmospheric chemistry processes into earth system models requires many **approximations and parametrizations**

$$\frac{\partial \chi(i)}{\partial t} = \text{Sources}(i) - \text{Sinks}(i) = E_i + C_i + A_i + T_i - W_i - D_i$$

- Considerations include: Emissions, Chemical mechanism, Aerosol model and cloud interactions, Transport, Dry Deposition, Wet Deposition
- Models allow us to perform multiple experiments regarding our atmosphere. Using the correct model and model configuration is important to correctly answer your question.